# THE UNITED REPUBLIC OF TANZANIA NATIONAL EXAMINATIONS COUNCIL OF TANZANIA

# 032 CHEMISTRY PRACTICALS FOR CERTIFICATE OF SECONDARY EDUCATION EXAMINATION, 2020

## A CHECKLIST OF CHEMICALS AND APPARATUSES

#### 1.0 IMPORTANT

The National Examinations Council has prepared a checklist of apparatuses and chemicals for Chemistry practicals. As a Head of the school, make sure that all the requirements indicated in this checklist are available in the school laboratory. Some of these will be used in the Certificate of Secondary Education Examination (CSEE) 2020 Chemistry practicals as will be prescribed in the 24 Hours Advance Instructions.

### 2.0 LIST OF APPARATUSES AND CHEMICALS

In addition to the normal fitting and reagents of CSE Chemistry laboratory, each candidate will require some of the listed apparatuses and chemicals as will be prescribed in the 24 hours advance instructions.

## 2.1 Apparatuses

- 1 burette (50 ml) ✓
- 2 boiling test tubes ✓
- test tube rack /
- 1 pipette filler ✓
- 1 pipette (20 ml or 25 ml)
- 4 conical flasks (250 ml)
- 1 conical flask cork/rubber stopper 🗸
- I retort stand and a clamp
- 1 tripod stand
- 1 beaker (50 ml)
- 1 beaker (100 ml)
- 5 beakers (250 or 300 ml)
- 1 wire gauze ~
- 1 pipette filler

- 1 measuring cylinder (10 ml)
- I glass/plastic funnel (Brim diameter which vranges from 50 mm to 100 mm)
- 1 measuring cylinder (50 ml) ~
- I measuring cylinder (100 ml)
- 1 stopwatch or stopclock
- I glass rod ✓
- 1 thermometer (0 °C−100 °C)
- 1 heat source for sharing in the ratio of 1:4 \( \sime \)
- 4 g cotton wool ~
- I water bath.
- 2 droppers (glass or plastic)
- 1 white tile ✓
- 1 filter paper ✓
- 2 both blue and red strips of litmus paper

### 2.2 Chemicals

- 1 g sulphuric acid ✓
- 4 g hydrated sodium thiosulphate.
- 3 g potassium permanganate 🗸
- 1 g hydrated oxalic acid ✓
- 1.5 g hydrochloric acid \square
- 2.5 g anhydrous sodium carbonate 🗸
- 500 cm<sup>3</sup> distilled water ✓
- 0.5 g phenolphthalein indicator \( \sqrt{} \)
- 0.5 g methyl orange indicator
- 4 g ammonium chloride

- 3 g sodium hydroxide ✓
- 0.4 g magnesium ribbon √
- 2.0 g magnesium carbonate
- 4 g zinc nitrate /
- 0.5 g silver nitrate ✓
- 2 g hydrogen peroxide
- I g nitric acid ✓
- 2 cm<sup>3</sup> ammonium solution
- 4 g iron (III) chloride
  - 4 g copper (II) nitrate

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- 2 g ammonium sulphate√
- 4 g calcium carbonate 🗸
- 4 g zinc chloride 🗸
- 4 g iron (II) sulphate
  - 4 g copper carbonate 🗸

- 4 g sodium chloride ✓
- 4 g copper (II) sulphate 4 g lead nitrate