043

SMZ

ZANZIBAR EXAMINATIONS COUNCIL

FORM THREE ENTRANCE EXAMINATION

CHEMISTRY

TIME: 2:30 Hours MONDAY 28TH NOVEMBER, 2016 AM

INSTRUCTIONS TO CANDIDATES

1. This paper consists of THREE (3) sections A, B and C.
2. Answer ALL questions in Section A and B, and any TWO (2) questions in section C. Question (9) is compulsory.
3. All answers must be written in the space provided.
4. Write your examination number on each page.
5. Cellular phones are not allowed in the examination room.
6. The following constants may be helpful.

H = 1, C = 12, Na = 23, O = 16

FOR EXAMINER'S USE ONLY		
QUESTION NUMBER	MARKS	SIGNATURE
1.		
2.		
3.		
4.		
5.		
6.		
7.		
8.		
9.		
10.		
11.		
TOTAL		

This paper consists of 10 printed pages

SECTION A: (30 Marks)

Answer ALL questions

1. Choose the letter of the best answer and write it in the table below.

i. The oxidation number of Carbon in Sodium hydrogen carbonate is

A) +2 B) 0 C) +1 D) +4

ii. Which of the following represents an anion?

A) NO_3^- B) CO_2 C) NH_4^+ D) SO_2

iii. The Nucleons are

A) Neutrons and Electrons C) Protons and Nuclides

B) Electrons and Protons D) Protons and Neutrons

iv. Pure water is

A) Acidic B) Neutral C) Basic D) Amphoteric

v. The symbols of the elements Iron, Zinc and Calcium are

A) Ir , Zn ,Ca B) Fe, Zi, Ka C) Fe, Zn , Ca D) Ir, Zn,Ka

vi. The valency of NO_3 in the compound $Al(NO_3)_3$ is

A) 1 B) 3 C) 9 D) 6

vii. The formula of Calcium hydrogen carbonate is

A) $CaHCO_3$ B) Ca_2HCO_3 C) $Ca(HCO_3)_2$ D) CaH_2CO_3

viii. The valency of an element with atomic number eleven (11) is

A) 2 B) 3 C) 0 D) 1

ix. The electronic configuration 2:8:2 represents that of the element

A) Potassium B) Magnesium C) Calcium D) Lithium

x. When oxygen reacts with Sulphur it will form

A) Basic oxide B) Neutral oxide

C) Amphoteric oxide D) Acidic oxide

Write the answers in this table

i	ii	iii	iv	v	vi	vii	viii	ix	x

2. Match the items in **List A** with the response in **List B**. Write the letter of the correct answer in the table below

List A	List B
i. Coal, charcoal, oil and gas.	A Class E fire
ii. It supports combustion.	B Flame
iii. Fire associated with electrical equipment.	C Rusting
iv. A chemical change occuring in iron or steel.	D Fire triangle
v. Oxygen, heat and fuel.	E Fuels
vi. Fire involving flammable liquids	F Sand and asbestos
vii. Coating of iron and steel with zinc.	G Class B fire
viii. Monoammonium Phoshate with a Nitrogen carrier.	H Luminous flame
ix. A team which put off fire when it is out of control.	I Painting
x. It uses Oxygen when burning but it produces soot.	J Class A,B and C fires
	K Fire squad
	L Galvanizing
	M Fire extinguisher
	N Non-luminous flame
	O Oxygen

Write the answers in this table

i	ii	iii	iv	v	vi	vii	viii	ix	x

3. Fill in the blanks. One word for each space.

i. A homogeneous mixture of _____ in the atmosphere is _____.

ii. The highest percentage of Nitrogen in the atmosphere is _____% by volume and that of Carbon dioxide is _____% by volume.

iii. White anhydrous Copper (II) Sulphate is used to test the presence of _____, which when present it turns anhydrous Copper (II) Sulphate to _____ hydrated.

iv. Oxygen reacts with non-metals to form _____ oxide and when it reacts with metals it forms _____ oxide.

v. The noble gases present in air include Helium, Krypton, Argon, _____ and _____.

Candidate's Examination Number _____

SECTION B: (50 Marks)

Answer ALL questions

4. a) Define the following terms

i. First Aid _____

_____

ii. First Aid Kit _____

_____

iii. Name any four (4) items that can be found in the First Aid Kit

_____

_____

b) i. Identify the situations that may require First Aid.

Vomitting Suffocating Crying Bleeding Choking Laughing
Scratching Sleeping Burning Walking Fainting Smiling

_____

_____

ii. Select any one (1) of the above situations in (b i.). Give the meaning.

_____

_____

_____

5. a) Differentiate between mass number and atomic number

_____

_____

_____

b) You are given an element with the symbol $^{12}_{6}X$

i. State the number of electrons present _____

ii. State the number of protons present _____

iii. State the number of neutrons present _____

iv. Sketch the energy shell diagram of that element X.

v. Name the element X _____

vi. Is it a metal or non - metal _____

6. a) Give the meaning of

i. Compound _____

_____

ii. Mixture _____

_____

iii. Solute _____

_____

b) Write down the methods that can be used to separate the following mixtures

i. Sand and water _____

ii. Iron filling and table salt _____

iii. Muddy water _____

iv. Salty water _____

c) Write one example in each of the following mixtures

i. Liquid - liquid mixture _____

ii. Solid - liquid mixture _____

iii. Solid - solid mixture _____

7. a) Define the following terms

i. A radical _____

_____

ii. Oxidation number _____

_____

b) Write the symbols of the following radicals

i. Hydroxyl _____ ii. Nitrate _____

iii. Sulphate _____ iv. Carbonate _____

c) By using the formula combine the metals Sodium , Calcium with the above radicals to form the compounds. Fill in the table below.

Elements	Radicals			
	Hydroxyl	Nitrate	Sulphate	Carbonate
Sodium (Na)				
Calcium (Ca)				

8. a) Define the term i. Flame _____

_____

ii. Heat _____

_____

b) Draw and label the bunsen burner.

c) What type of flame will occur when the air holes of the Bunsen burner

i. Are fully opened _____

_____

ii. Are closed _____

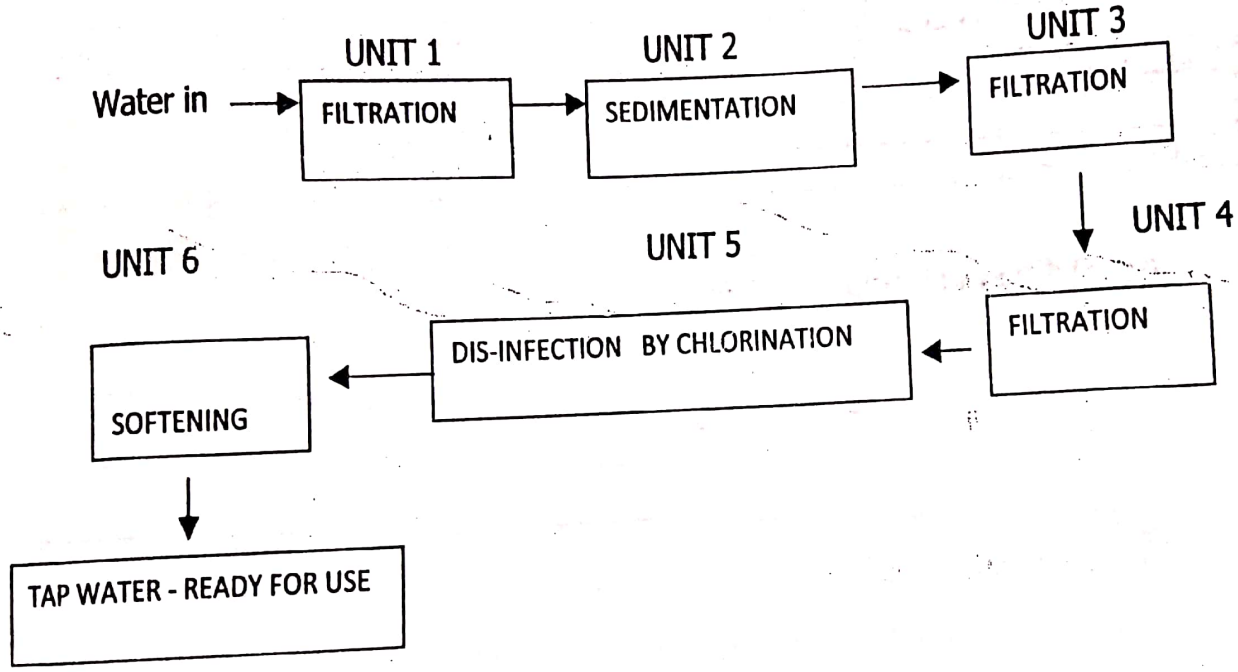
_____

SECTION C: (20 Marks)

Answer any TWO (2) questions

Question 9 is compulsory, answer either (9a) or (9b)

9. a) Study the chart below which shows the water treatment, then answer the questions.



i. List two methods of separation of mixtures that are shown in the chart

_____ and _____

ii. Explain the meaning of the water treatment

_____

_____

_____

iii. Mention a common local method which is used for domestic water purification

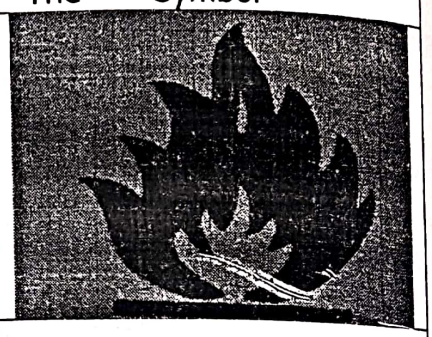
_____

iv. From the chart name the unit which shows water is treated with chemical_____

v. Name that chemical _____

vi. Write the symbol of that chemical_____

9. b) A student is asked to put labels on bottles. The labels should contain warning sign and symbols. Fill in the table below. The first one is done as an example.

	The bottles containing	Warning sign	The Symbol
1.	Spirit	Flammable	
2.	Conc. Sulphuric acid		
3.	Petrol		
4.	Mercury		

a) Define the term energy.

b) Identify with examples three (3) categories of fuels

c) "The use of charcoal and fire wood has greater environmental effects" Advise the villagers in your community on

i. Disadvantage of deforestation.

ii. Use of alternative sources of fuel

iii. The preventive measures taken for environmental conservation.

11. a) Define the term

i. Isotope_____

_____

ii. Name the two (2) isotopes of chlorine

_____

b) Write any three (3) assumptions of Dalton's Atomic theory.

_____

_____

_____

_____

c) List three (3) properties of the neutron

_____

_____

_____

_____

d) From $N=2n^2$

Find the maximum number of electrons (N), when n = 2, and n = 3.

_____

_____

_____

_____

_____

_____
